Redox Behavior and Radical Scavenging Capacity of Hepatoprotective Nutraceutical Preparations

Douglas Vieira Thomaz¹, Pierre Alexandre dos Santos¹

¹ Federal University of Goiás, Faculty of Pharmacy. 240 st., Leste Universitário Sector, Goiânia, Goiás State, ZIP: 74605-170, Brazil

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Correspondence: Address correspondence to this author at the Federal University of Goiás, Faculty of

Pharmacy. 240 st., Leste Universitário Sector, Goiânia, Goiás State, ZIP: 74605-170, Brazil;

Email: douglasvthomaz@gmail.com

Background: Over-the-counter hepatoprotective nutraceuticals are highly commercialized preparations worldwide. However, their alleged antioxidant capacity and health benefits are still not fully understood.

Objective: This work showcased the first investigation of the redox behavior of hepatoprotective nutraceuticals by spectrophotometric and electrochemical approaches.

Method: The samples were segregated into two groups, namely: A, B, and C based on isolated compounds (IC); and D, E and F based on standardized herbal extracts (SHE).

Results: Results evidenced that IC showcase similar response and distinctions could be attributed to varying concentrations of choline. In SHE, the slopes showcased superimposition due to the presence of *Peumus boldus*. The electrochemical assays showcased that samples A and C exhibited a single anodic peak at Ep1a $\approx +0.7$ V, which could be attributed to the oxidation of methionine; while samples D, E and F, showcased two anodic peaks at Ep1a $\approx +0.35$ V and Ep2a $\approx +0.7$ V, suggesting the oxidation of phenolic and amine moieties respectively.

Furthermore, the first two principal components explained 84.8% of all variance in the model, thereby suggesting statistical reproducibility.

Conclusion: This work showcased the first investigation of the redox behavior of hepatoprotective nutraceuticals, thereby shedding light on their antioxidant capacity and physical-chemistry.

Keywords: Thermodynamics, kinetics, antioxidant capacity, natural products, free radical, hepatoprotective nutraceuticals.